

**GUJARAT TECHNOLOGICAL UNIVERSITY**  
**B.Ph. – SEMESTER- III EXAMINATION – WINTER -2021**

**Subject Code: BP302TP****Date: 17/02/2022****Subject Name: Physical Pharmaceutics - I****Time: 10:30AM to 01:30PM****Total Marks: 80****Instructions:**

1. Attempt any five questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.

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|-------------|-----|---|-----------|
| <b>Q.1</b>  | (a) | Describe the solute - solvent interactions that influence the solubility of drugs in liquids.   | <b>06</b> |
|             | (b) | Define & explain solubility. Discuss different solubility expressions.  | <b>05</b> |
|             | (c) | Explain Henry's law and factors affecting for solubility of gases in liquids.   | <b>05</b> |
| <b>Q.2</b>  | (a) | Define & explain for the state of matter (i) Critical point (ii) Eutectic mixture.  | <b>06</b> |
|             | (b) | Describe different types and applications of liquid crystals.   | <b>05</b> |
|             | (c) | Write short note on polymorphism.   | <b>05</b> |
| <b>Q.3</b>  | (a) | Define surface tension and express it in terms of surface free energy.  | <b>06</b> |
|             | (b) | Describe capillary rise method to determine surface tension.  | <b>05</b> |
|             | (c) | Write short note on spreading co-efficient.   | <b>05</b> |
| <b>Q.4</b>  | (a) | Define complexes. Write applications of complexation in pharmacy.   | <b>06</b> |
|             | (b) | Write short note on chelate type complexes.   | <b>05</b> |
|             | (c) | Describe solubility method to determine the formation of complex and its stability constant.  | <b>05</b> |
| <b>Q.5</b>  | (a) | Explain buffer & buffer capacity. Write applications of buffers in pharmacy.  | <b>06</b> |
|             | (b) | Describe electrometric method to determine pH.  | <b>05</b> |
|             | (c) | Explain isotonic solutions & paratonic solutions. Describe freezing point depression method for adjusting the tonicity of a solution. | <b>05</b> |
| <b>Q. 6</b> | (a) | Explain distribution law. Write its limitations and applications.   | <b>06</b> |
|             | (b) | Explain Refractive Index. Write its applications.   | <b>05</b> |
|             | (c) | Define Dielectric Constant. How it is measured?   | <b>05</b> |
| <b>Q.7</b>  | (a) | Write short note on Hydrophilic - Lipophilic Balance (HLB).   | <b>06</b> |
|             | (b) | Write short note on Langmuir adsorption isotherm.   | <b>05</b> |
|             | (c) | Explain protein binding. Write importance of protein binding.   | <b>05</b> |

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