|           | Shree H.N.Shukla group of colle<br>PHYSICS<br>T.Y.B.Sc. (Sem. VI) (CBCS)<br>Unit Test examination<br>PAPER- 602<br>UNIT - 2 : CRYSTAL STRUCTURE & BINDI                                             |                    |
|-----------|-----------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--------------------|
| DATE :    | Total marks : 30                                                                                                                                                                                    |                    |
| Instructi | ons: All questions are compulsory.<br>The right side figure indicates total marks of the q<br>Draw the figure wherever necessary.<br>Write answers of all the questions in main answer<br>SECTION-A |                    |
| Q.1: One  | e marks questions:                                                                                                                                                                                  | [5 MARKS ]         |
| 1         | Define : Basis, Lattice                                                                                                                                                                             |                    |
| 2         | Give the names of seven crystal system.                                                                                                                                                             |                    |
| 3         | Define : Hydrogen bond.                                                                                                                                                                             |                    |
| 4         | In covalent bond the spins of two electrons are                                                                                                                                                     | ·                  |
| 5         | Give one example of ionic crystal and covalent bond                                                                                                                                                 |                    |
|           | SECTION – B                                                                                                                                                                                         |                    |
| Q.2 (A):  | Short Questions: Write any three : [2 Marks each]                                                                                                                                                   | [6 MARKS]          |
| 1         | Draw the plane of miller indices (001), (010).                                                                                                                                                      |                    |
| 2         | Draw the plane of miller indices (011), (111).                                                                                                                                                      |                    |
| 3         | Find the ration of interceptson the three axes by (12 cubic crystal.                                                                                                                                | 3) plane in a simp |
| Q.2 (B) : | Short questions: Write all three: [3 Marks each]                                                                                                                                                    | [9 MARKS]          |
| 1         | Explain simple cubic (SC) structure.                                                                                                                                                                |                    |
| 2         | Explain inonic crystal bond.                                                                                                                                                                        |                    |
| 3         | Explain covalent crystal bond .                                                                                                                                                                     |                    |
| Q.2 (C):  | Write Detail Note on [Any two]: [5 Marks each]                                                                                                                                                      | [10 MARKS]         |
| 1         | Describe : Miller indices in detail.                                                                                                                                                                |                    |
| 2         | Explain : Hexagonal closed packed structure.                                                                                                                                                        |                    |
| 3         | Explain : metalic bond.                                                                                                                                                                             |                    |
| 4         | Explain : BCC crystal structure.                                                                                                                                                                    |                    |