Shree H.N.Shukla group of colleges			
PHYSICS			
PAPER- 601			
SECTION-A			
Q.1: One marks questions:			[5 MARKS]
1	The ionisation power of $lpha$ particle is time greater than eta rays.		
2	The particle are identical with electron.		
3	The unit of radioactivity is		
4	The $\boldsymbol{\beta}$ particle have charges.		
5	Write equation of Half life time :		
6	The ionisation power of $lpha$ particle is time greater than eta rays.		
7	The particle are identical with electron.		
8	The unit of radioactivity is		
9	The $oldsymbol{eta}$ particle have charges.		
10	Write equation of Half life time :		
11	The ionisation power of $lpha$ particle is time greater than eta rays.		
12	The particle are identical with electron.		
13	The unit of radioactivity is		
14	The $oldsymbol{eta}$ particle have charges.		
15	Write equation of Half life time :		
SECTION – B			
Q.2 (A): Short Questions:		[2 Marks each]	
1	What is the mass of 1 C	urie of U ²³⁸	
2	A radioactive substance has half life of 30 days . Claculate the radioacitve disintegraiton constant.		
3	A radioactive substance has half life of 50 days . Find the radioactive decay constant and mean llife.		
4	Write two reaction of (α , p).		
5	Write two reaction of (n , Υ).		
6	Write any two reaction by alpha particles.		

7 What are called magic numbers ? 8 Binding energy of SC is 424.326 MeV find binding energy per nucleaon. Explain : Mirror Nuclei. 9 Q.2 (B) : Short questions: [3 Marks each] Write properties of APLHA rays. 1 Write properties of GAMMA rays. 2 3 Explain the radioactive THORIUM series. Discuss : Binding energy. 4 5 Explain : Nuclear size. 6 Define : Nuclear density. 7 Explain the working of solid state detector. 8 Describe the characterisitics of GM counter. Explain Photomultiplier. 9 Q.2 (C): Write Detail Note ON : [5 Marks each] 1 Explain Half life and Mean life . 2 Explain the application of Radio isotope. Describe the theory of ALPHA decay. 3 Explain natural radioactivity series. 4 5 Describe the interaction between energetic particle and matter. Obtain the Q value for nuclear reaction. 6 Explain nuclear transmuation. 7 Explain conservation laws in nuclear reaction. 8 9 Describe Rutherford alpha scattering experiment. Describe classification of nuclei. 10 Explain qualitative facts about size , mass and charge of nucleas. 11 Write semiempirical mass formullaznd explain its terms. 12