

# Shree H.N.Shukla Group of Colleges <u>B</u>. Sc. (Sem-VI) (CBCS) (Unit Test-I)

C-602: Organic and Spectroscopy

Time: 1.5 hours

Date:10/04/2021 Total Marks: 30

#### **Instructions**

1. All Question are compulsory.

(ii) Explain Isoelecric point (PI).

2. Figure to the right indicate the full marks of Questions.

## 

### (D) Answer the Following (any one).

[05]

- (i) Give the reaction due to both the amino (-NH<sub>2</sub>) and carboxylic (-COOH) group in  $\alpha$ -amino acid.
- (ii) Prove the structure of Thyroxine.

### Q.2 (A) Answer the Following.

[01]

- (i) What is full form of NMR and TMS.
- (ii) Which isotope of Carbon in NMR active?
- (iii) Give Number of signal in the compound given below.

- (iv)Calculate DBE for given molecular formula C<sub>8</sub>H<sub>8</sub>O<sub>2</sub>
- (v) Write the examples of nuclei, which possess spin value(I)=1/2

### (B) Answer the Following (any one).

[02]

- (i) Define: geminal and vicinal coupling.
- (ii) Why TMS used as reference in NMR spectroscopy.

### (C) Answer the Following (any one).

[03]

- (i) Explain shielding and deshielding effect.
- (ii) Determine the molecular structure from the following data:

 $M.F.:C_8H_8O_2\\$ 

IR :  $3400_{(br)}$ , 2950, 2530,1715, 1300, 1060, 810 cm<sup>-1</sup>

NMR:

No. of Signal	No. of	Multiplicity	Chemical
	protons		$shift(\delta_{ppm})$
а	3H	S	2.01
b	1H	S	10.75
С	4H	m	7.15

## (D) Answer the Following (any one).

[05]

(i) Determine the molecular structure from the following data:

 $M.F.\ C_9H_{10}O_2$ 

IR: 3030, 2930, 1670, 1598, 1258, 1021, 833 cm<sup>-1</sup>

NMR:

No. of Signal	No. of protons	Multiplicity	Chemical shift( $\delta_{ppm}$ )
а	3H	S	2.50
b	3H	S	3.90
С	4H	complex	7.83

(ii) Write a short note on Deuterium labelling.