

SHREE H. N. SHUKLACOLLEGE OF SCIENCE

(AFFILIATED TO SAURASHTRA UNIVERSITY) Nr. Lalpari lake, Behind old Marketing Yard, Amargadh, Rajkot-360001, Ph. No-9727753360

SAURASHTRA UNIVERSITY, RAJKOT SYLLABUS FOR BIOCHEMISTRY SEMESTER – II (NEP-2020)

Semester-II

Course Category	MDC/IDC-1
	In addition to courses mentioned in SOP basket;
	Recommended for Physical Science, Mathematical
	Science, Life science Programs
Title of the Course	Biochemistry:Introduction to Biomolecule - I
Course Credit	03
Teaching Hours per Sem.	45
Total Marks	75

 iii I: Water and Chemical Bonds: Water: Essentiality to life. Water as a biological fluid. Special properties of water. Hypotonic, hypertonic and isotonic solutions. Effects of osmotic pressure on living cells. Chemical Bonds: Types of bonds and Bond energy Major types of chemical bonds -, Covalent bond, Ionic bond Resonance Bond Metallic 		Course Content	Hours
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bond, Hydrogen bond and Van der waal forces and their significance	• • •	 Water: Essentiality to life. Water as a biological fluid. Special properties of water. Hypotonic, hypertonic and isotonic solutions. Effects of osmotic pressure on living cells. Chemical Bonds: Types of bonds and Bond energy Major types of chemical bonds -, Covalent bond, Ionic bond Resonance Bond Metallic bond, Hydrogen bond and Van der waal forces and their significance 	9 hrs.

Unit II: Carbohydrates I

Carbohydrates I: Classification and biological importance Monosaccharides:

- Configuration relationship of D-aldoses, D-ketoses.
- Reactions of glucose and fructose- oxidation, reduction, reducing properties, formation of glycosides, acylation, methylation, condensation phenyl hydrazine, addition HCN.
- Stereochemistry of monosaccharides, (+) and (-), D and L, epimers, anomers, enantiomers and diastereomers.
- Glucose: Elucidation of open chain structure, configuration and ring structure of glucose and mutarotation. Open and Haworth structures of galactose, mannose, ribose and fructose.
- Structure and biological importance of amino sugars, deoxy sugars, sugar acids, neuraminic and muramic acid.

Unit III: Carbohydrates II

Disaccharides: Structure and Importance

- Maltose, isomaltose, lactose, Sucrose, cellobiose, trehalose and Invert sugar.
- Polysaccharides: Classification with examples.
- Homopolysaccharides: Partial structure, occurrence and importance of starch, glycogen, inulin, cellulose, chitin, and pectin.
- Heteropolysaccharides: Occurrence, importance and the structure of the repeating units of Glycosaminoglycans- heparin, hyaluronic acid, and chondroitin sulphate.
- Blood group oligosaccharides.
- Chemical basis of the qualitative tests: Molisch, iodine, Benedicts, Fehling's, picric acid, Barfoed's, Bial's, Seliwanoff's, osazone tests.

Unit : IV Simple Lipids

9 hrs.

9 hrs.

9 hrs.

• Lipids: Definition, classification and biological role,	
• Fatty acids: Saturated [C4-C24] and unsaturated fa	atty acids: Nomenclature, structure
&occurrence. Physical properties and chemical reacti	ons: esterification and rancidity.
• Essential fatty acids: (ω -3 & ω -6 fatty acids): stru	cture, occurrence &
biological importance	
• Tri-acylglycerols: simple and mixed glyceride	s with examples,
Saponification, hydrolysis, Definition & significance	of saponification value, iodine
value, acid value and per-oxide value.	
• Waxes: Composition, importance with examples	
Unit : V Compound Lipids	9 hrs.
• Phosphoglycerides: Structure of lecithin, cephalins,	phosphotidyl
inositol, plasmalogens, and cardiolipins. biological ro	le of
phosphoglycerides.	
• Sphingolipids : Ceramides, structure and importance	of sphingomyelin.
Glycosphingolipids : Structure and importance of	cerebrosides
(galactocerebroside andglucocerebroside), gangliosi	des (GM1, GM2, GM3).
• Eicosanoids: Structure of PGE1, PGE2, PGF1 α and	$1 \text{ PGF}_{2\alpha}$. Biological
roles of thromboxanes, leukotrienes and prostagland	ins.
• Plasma lipoproteins: Types and functions, compositi	on and structure of lipoprotein.

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6	In addition to courses mentioned in SOP basket;	
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	science Programs	
Title of the Course	Biochemistry -P: Introduction to Biomolecule - I Practical	
Course Credit	01	
Teaching Hours per Sem.	30	
Total Marks	25	

Suggested laboratory experiments:

- 1. Safety measures and introduction to the instruments used in biochemistry laboratory
- 2. Importance of calibration of instruments and cleaning of glassware
- 3. Qualitative tests of monosaccharides.
- 4. Qualitative tests of Disaccharides and Polysaccharides.
- 5. Estimation of reducing sugar by DNSA method
- 6. Isolation of starch from potato.
- 7. Acid Hydrolysis of Starch.
- 8. Qualitative tests for lipids.
- 9. Determination of Acid Value of oils.